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them with their charge 2 draw six neutrons in the nucleus of  
the atom 3 draw two electrons in the first energy level and  
label them with their charge 4 draw three electrons in the  
second energy level and label them with their charge 5 what  
element is represented by the diagram

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atom atoms have protons and neutrons in the center making  
the nucleus while the electrons orbit the nucleus the modern  
atomic theory states that atoms of one element are the same  
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web a carbon 12 12 c b this atom contains six protons and six  
neutrons there are six electrons in a neutral 12 c atom the  
net charge of such a neutral atom is zero and the mass  
number is 12 c the preceding answers are correct d the atom  
will be stable since c 12 is a stable isotope of carbon e the  
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 web the atomic number tells you the number of of an n one atom of an element it also tells you the number of in a neutral atom of that element the atomic number gives the identity of an element as well as ts location on the periodic table no two different elements will have the give the symbol and number of protons in one atom of atomic number

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web a both the electron geometry and the molecular structure are octahedral b both the electron geometry and the molecular structure are trigonal bipyramid c both the electron geometry and the molecular structure are linear d both the electron geometry and the molecular structure are trigonal planar

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