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sulfur dichloride n<sub>2</sub>o<sub>4</sub> dinitrogen

tetroxide the a of tetra is dropped

before the o of oxide p<sub>4</sub>o<sub>10</sub>

tetraphosphorous decoxide the a of

deca is dropped before the o of oxide

pf<sub>5</sub> phosphorous pentafluoride

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h which immediately combines with a

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h<sub>3</sub>o<sup>+</sup> the h<sub>3</sub>o<sup>+</sup> in is what defines the

acidic properties of a substance

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to form the hydronium ion h<sub>3</sub>o<sup>+</sup> the h<sub>3</sub>o<sup>+</sup>

ion is what defines the acidic properties

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in model 2 that contain sulfur

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base in the proton h<sup>+</sup> ion transfer that

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