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web understand that parts of an acid base titration be able to determine the k_a or k_b from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate k_{a1} and k_{a2} for a polyprotic acid by the end of this lab students should be able to

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was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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hydrochloric acid and the base is naoh sodium hydroxide when the acid and base react they form nacl sodium chloride which is also known as table salt

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