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web answer the correct option is c explanation the molarity of a mixture mmix can be calculated using the following formula $m_{\text{mix}} = \frac{m_1 + m_2}{V_1 + V_2}$ $\frac{1.5 \times 480 + 1.2 \times 520}{480 + 520} = 1.344 \text{ M}$ q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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6.1 calculating molarity problems

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solution 0 2074 g of calcium hydroxide
ca oh 2 in 40 00 ml of solution 10 5 kg
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web molality differs from molarity only
in the denominator while molarity is
based on the liters of solution molality

is based on the kilograms of solvent
concentrations expressed in molality
are used when studying properties of
solutions related to vapor pressure and
temperature changes

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show all work including units watch
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molarity is one of the many ways to
measure concentration or