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initial volume v₁ 10 liter 10 dm³ 10 x
10 3 m³

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kinetic energy of the ice goes up
resulting in a higher average speed and
thus an average greater position
variance of the molecules in the ice the
reservoir does become more ordered
but due to its much larger amount of
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 $\Delta w = d p v$ constants solution in an
isothermal process temperature
remains constant $\Delta t = 0$ since internal
energy depends on the temperature Δu
0 from first law of thermodynamics Δu
 $\Delta q = \Delta w$ since $\Delta u = 0$ $\Delta q = \Delta w$ also $p v = n r t$ as t
is constant $p v$

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