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## M Mark

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cycle is divided in 2 phases interphase consists of 3 steps s phase g2 phase g phase consists of 5 phases prophase prometaphase metaphase anaphase telophase during this stage mitotic spindle begins to form chromosomes start to condense in this stage nuclear envelope break down

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0 chromosome 11 interphase 4 the cell  
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interphase m mitosis g 0 gap 0 g 1 gap  
1 g 2 gap 2 s synthesis g 3 gap 3 cell  
division is the process by which a  
parent cell divides into two daughter

cells 1 cell division usually occurs as  
part of a larger cell cycle in which the  
cell grows and replicates its  
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cell division there are two faces named  
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a non polar atom is fat dissolvable it s  
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as is ethanol when you blend water and  
ethanol you see a homogeneous  
arrangement with the two

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web q1 arrange the bonds in each of the following sets in order of increasing polarity  $\text{c-f } \text{be-f } \text{o-f } \text{o-cl } \text{s-br } \text{c-p } \text{c-s } \text{b-f } \text{n-o}$  q2 determine based on electronegativity and symmetry if the following molecules are polar or nonpolar  $\text{hcn } \text{h-c-n } \text{cf}_4 \text{ c-f}_4 \text{ cocl}_2 \text{ c-o-c-l}_2 \text{ nf}_3 \text{ n-f}_3$  q3

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web dec 2 2016 a polar bond is one where the charge distribution between the two atoms in the bond is unequal a polar molecule is one where the charge distribution around the molecule is not symmetric it results from having polar bonds and also a molecular structure where the bond polarities do not cancel

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web when the atoms in a bond are not the same the bonding electrons are shared and the bond is

electronegativities equally unequally non polar polar unequally polar the degree of polarity of a bond between any two atoms is determined by consulting a table of

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web chemical bonds hold molecules together and create temporary connections that are essential to life types of chemical bonds including covalent ionic and hydrogen bonds and london dispersion forces introduction living things are made up of atoms but in most cases those atoms aren't just floating around individually

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web explain how a molecule that contains polar bonds can be nonpolar answer problem 6 2 2 6 2 2 which of the following molecules and ions contain polar bonds which of these molecules and ions have dipole moments a clf 5 b clo 2 clo 2 c tecl<sub>2</sub> 4 tecl<sub>4</sub> 2 d pcl<sub>3</sub> e sef<sub>4</sub> f ph<sub>2</sub> ph<sub>2</sub> g xef<sub>2</sub> 2 answer problem 6 2 3 6 2 3

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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl

end of the molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as bf<sub>3</sub> trigonal planar ch<sub>4</sub> tetrahedral pf<sub>5</sub> trigonal bipyramidal and sf<sub>6</sub>

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web explain the difference between a nonpolar covalent bond a polar covalent bond and an ionic bond answer nonpolar covalent electronegativity difference is less than 0 4 nonmetal nonmetal close together on the periodic table

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web apr 1 2021 updated on april 01 2021 chemical bonds may be classified as being either polar or nonpolar the difference is how the electrons in the bond are arranged key takeaways what is a polar bond in chemistry a polar bond is a type of covalent bond in which the electrons forming the bond are unequally distributed