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web what is the mole fraction of LiCl in the solution 11.8 practice problem acetone C₃H₆O has a density of 0.791 g/mL and xylene C₈H₁₀ has a density of 0.864 g/mL a solution was prepared by mixing acetone and xylene calculate the mole fraction of acetone in a solution if 19.0 mL of acetone was dissolved in 98.4 mL of xylene 7 1
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web since the mole fraction of the O₂ is 0.050 it contributes 0.050 or if you will 5.0% of the total pressure thus total pressure is 0.210 + 0.050 = 0.260 atm problem 9 a sample of 1.43 g of helium and an unweighed quantity of O₂ are mixed in a flask at room temperature