

# Bates Physical Exam

**Michal Rosen-Zvi**

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98 use the average atomic mass that

you calculated in lab 2 see the  
background information section at the  
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neutrons each do

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atoms of the same element with the  
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but different masses due to different  
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atomic mass of an element is the  
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most people respond that you add up  
the values and divide by how many  
there were the average of 2 2 4 and 8 is  
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web procedure 1 obtain a sample of  
candium 2 separate the three isotopes  
and count obtain the total mass for  
each group of isotopes 3 complete the  
data table using the instructions found  
under the data analysis questions  
section the large x s in the table  
indicate cells that do not require data

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web candium is an interesting element  
found only at fordham high school for  
the arts we will be using this special  
new element today to learn about  
atomic mass calculations the atomic  
mass listed on the periodic table for  
each element is a weighted average  
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web 1 obtain a sample of candium  
separate the 3 isotopes candium red  
hots mini candium peanut butter  
regular and candium peanut large and  
record the numbers for each type of

isotope 2 next determine the mass for  
each isotope and calculate the total  
mass in grams 3 calculate the average  
mass

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web cole quinn period 5 mr hans  
honors chemistry candium post lab  
questions 1 no the average atomic mass  
of my group s candium sample was very  
similar to our neighbors samples if each  
group were to have used larger  
samples of candium the respective  
differences would have been smaller  
because the larger the sample size the  
greater the

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web obtain a sample of candium from  
your teacher 2 separate the three  
isotopes m m s skittle s and peanut m m  
s and measure the mass of each isotope  
record in data table 1 1 3 count the  
number of m m s skittle s and peanut m  
m s record in data table 1 1

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web procedure 1 place the candium  
atoms in a paper towel or a small bag 2  
gently shake the atoms for 30 seconds  
this is the estimated half life of your  
candy 3 gently pour out the atoms onto  
a paper towel 4 count the number of  
atoms with the print side up these  
atoms have decayed 5

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electronic balance safety when handled  
correctly candium has no known safety  
hazards candium samples should  
always be placed in a paper cup for all  
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paper to avoid any contact with the  
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mass of a fictitious element called  
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