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Michael Seilmaier

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molecule form a trigonal planar  
geometry all angles 120 degrees four  
regions of electron density around a  
central atom in a molecule form a  
tetrahedral geometry all angles 109 5  
degrees

## **molecular geometry and polarity phet tiss** - Mar 11 2023

web in this investigation you will  
examine a theory that chemists use to  
explain different aspects of chemical  
bonding valence shell electron pair  
repulsion vsepr theory attention will be

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given to how molecules are arranged in different shapes and how chemists can molecule  
predict the geometry of a given