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Peipei Pang

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obvious when you look at a chemical equation for a reaction when you see a chemical equation in a textbook it looks like the reactants are completely used up and all made into product

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web jun 2 2020 the reactant that is consumed first and limits the amount of product s that can be obtained is the limiting reactant to identify the limiting reactant calculate the number of moles of each reactant present and compare this ratio to the mole ratio of the reactants in the balanced chemical equation

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in the real world amounts of reactants and products are typically measured by mass or by volume it is first necessary to convert the given quantities of each reactant to moles in order to identify the limiting reactant

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reactant it doesn't matter which

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