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Lei Shi

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radioactive decay of a nucleus is too
small to see with the naked eye we can
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radioactive element can be modelled
using cubes dice or coins in decay a
radioactive parent nucleus randomly
emits an alpha or beta particle
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of radioactive decay the probability p
that a certain atom will decay in a small
amount of time t is $p = \lambda t$ $1 - \lambda t$ $\ln 2$ $t = \frac{1}{\lambda}$ 2 where λ

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emits a helium nucleus with 2 neutrons
and 2 protons thus the atomic mass of
the new element is decreased by four
and the atomic number

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web nov 21 2021 the equation below describes the radioactive decay of a uranium 238 atom where a thorium 234 atom and an alpha particle are formed
 ${}^{92}_{238}\text{U} \rightarrow {}^{90}_{234}\text{Th} + {}^2_4\text{He}$
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