

The Rat Fat Unity

Jianjun Gao

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introduction for this test i had five

containers labelled a e each containing a metal salt i had to find out which metal is used in each of the five unknown salts by using a metal loop placing it in the unknown salts and testing it by putting the metal under a roaring bunsen flame safety points wear googols all the time when

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absorption emission spectroscopy flame test lab ldc 5 1 atomic emission the flame test when a small amount of a solution of a metal ion is placed in the flame of a bunsen burner the flame turns a

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web task use a flame test to determine which ion salt produces the red color then using a spectroscope match the bright line spectra from red fireworks to a substance in your material list then make your final determination of what ion in the material list causes red fireworks procedure

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web oct 15 2022 the objectives of this lab are to perform flame tests of metal

cations in order to observe their characteristic colors perform calculations for determines the output and energy of the emitted

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web energy and matter answered worksheet the experiment conducted aimed to give students the opportunity to determine different metal ions in liquid and solid compounds by observing the color of the flames they give off when

8 flame tests of metal cations

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web sep 22 2021 pre laboratory assignment flame tests of metal cations in this lab you will perform flame tests of several different metal cations the characteristic colors observed are due to emitted electromagnetic radiation from the excited metal cations in this lab how do the metal cations become excited

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web feb 10 2021 section 1 purpose and summary observe the color of light emitted when a sample of a salt of a metal cation is heated use the

characteristic color of flames to identify ions in various laboratory samples observe the emission spectra of various elements in gas discharge tubes

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web 25 october 2022 objective experiment 1 to show how flame test can be used to identify some metals ion 2 to identify some unknown metal ions using flame test introduction the arrangement of electrons around the nucleus of the atom is

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