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this section you should be able to  
answer these questions what affects  
how a gas behaves what are the gas  
laws what affects the behavior of a gas  
gases behave differently than solids or  
liquids gas particles have a large  
amount of space between them

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the volume of its 101 300 pa a pressure  
of 101 3 kpa is equal to the air inside a  
fully pumped basketball has a higher  
pressure than the air outside because  
there is a higher number of gas  
particles per unit volume does the air  
inside a fully pumped basketball have a  
higher or lower pressure than the air  
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answer each question on the lines  
provided you must include the terms  
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which variable is held constant in the  
relationship described by boyle s law 2  
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fills measured in cubic centimeters cm<sup>3</sup>  
milliliters ml liters l gas particles move  
and fill the space available this  
measure of gas is the same as the  
measure of its container a measure of  
the average energy of random motion  
of the particles of a substance  
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to fill its container unlike solids or  
liquids the reverse is also true they are  
easily compressed or squeezed into a  
smaller volume compressibility is a  
measure of how much the volume of  
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fluids have no definite shape or volume  
and they expand to fill their container  
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molecules are in constant motion and  
collide with each other and walls of the  
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are far apart are compressible

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 your expectations of people third  
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