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problems using the following equation

2 nh₃ + g h₂ so 4 nh₄ 2 so₄ aq

calculate the volume of nh₃ in liters

needed at 20°C and 25 atm to react

with 150 kg of h₂ so 4

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84.2 carbon by mass and has a density

of 0.8205 g/ml determine the mass of

carbon dioxide produced during a 500

mile trip 3 785 liters per gallon answer

1 28 10 5 g co₂

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please answer the following on

separate paper using proper units and

showing all work please note that these

problems require a balanced chemical

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if 1.0 l of carbon monoxide reacts with

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10 17 o₂ 12 co₂ using the following

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born on march 24 1886 in highland

park illinois as a child he saved pennies

to buy used photographic equipment at

the age of 16 his father gave him a

kodak bulls eye number two camera

weston began to take pictures at his

aunt s farm and in the parks of chicago

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king s college cambridge where he was
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