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$$\text{i}2\text{o}5 + 5\text{co} \rightarrow \text{i}2\text{o}5 + 5\text{co}2$$

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$$2\text{sr} + \text{o}2 \rightarrow 2\text{sro}$$

$$\text{sr}0 \rightarrow \text{sr}2 \text{ oxidized}$$

$$\text{o}0 \rightarrow \text{o}2 \text{ reduced}$$

$$\text{ox ag } 2\text{li} + \text{s} \rightarrow \text{li}2\text{s}$$

$$\text{li}0 \rightarrow \text{li}1 \text{ oxidized}$$

$$\text{red ag } \text{s}0 \rightarrow \text{s}2 \text{ reduced}$$

$$\text{ox ag } 3\text{cs} + \text{br}2$$

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