

Lakshmi Stotram In Telugu

N Colangelo

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 $\text{NaBr} + \text{CaOH}_2 \rightarrow \text{CaBr}_2 + \text{NaOH} + \text{NH}_3$
 $\text{H}_2\text{SO}_4 + \text{NH}_4\text{OH} \rightarrow \text{NH}_4\text{SO}_4 + \text{H}_2\text{O}$
 $\text{H}_3\text{PO}_4 + \text{H}_2\text{PO}_4 \rightarrow \text{Li}_3\text{N} + \text{NH}_4\text{NO}_3 + \text{LiNO}_3$

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included and is available for download some additional information to be

aware of for teachers demonstrations 1 and 5 will not produce a chemical

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1 Na_3PO_4 3 koh 3 naoh K_3PO_4 double displacement 2 MgCl_2 Li_2CO_3 MgCO_3 2

LiCl double displacement 3 C_6H_{12} 9 O_2 6 CO_2 6 H_2O combustion 4 pb feso 4 PbSO_4

fe single displacement 5 CaCO_3 CaO CO_2 decomposition 6 P_4 3 O_2 2 P_2O_3

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2AgCl 3 2 LiBr f 2 2 LiF br 2 4 2 na f 2 5 f 2 Br_2 b balance the chemical reactions

and identify each type

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oh 2 Al_2SO_4 3 CaSO_4 2al oh 3 type of

reaction double displacement 3 $3\text{Mg} + 1\text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$ type of reaction single displacement 4 $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$ type of reaction combustion 5 $2\text{PbSO}_4 + 2\text{PbSO}_3 + \text{O}_2$ type of reaction

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which one or more substances are converted to one or more different substances the starting substances are called the reactants and the new substances that form are called the products

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