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mass of one proton mass of one neutron
a sample of cesium cs has the following
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134 4 7 weighted average atomic mass
132 x 0 200 133 x 0 752 134 x 0 047
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purpose what is an isotope what does it
mean to say that the atoms in a sample
of an element are isotopes of each
other ordinary beans are a lot bigger
than atoms but perhaps they can give
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have talked in class about isotopes
atoms of the same element that differ in
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lab questions define average atomic
mass write a mathematical equation
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web beanium isotope b brownium
beanium isotope c whiteium of atoms
beans mass beans in cup mass empty
cup mass of atoms beans 2 minus 3
average mass of 1 bean 4 divided by 1
of beans 1 a b c x 100 section 3
calculate the average atomic mass of
beanium

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purpose this lab exercise is designed to
show you isotopes of an element in a
simulation form you will be asked to
gather data about the isotopes and
organize the data if atoms were as
large as beans they could be sorted
counted and massed

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web in the beanium lab students obtained a sample of beanium a fictitious element in order to learn about isotopes and average atomic mass beanium has three isotopes white brown and speckled which have different masses and different percent abundances

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web your final assignment is to calculate the average atomic mass of beanium to be placed on the periodic table the formula is abundance of isotope 1 x mass of isotope 1 abundance of isotope 2 x mass of isotope 2 abundance of isotope 3 x mass of isotope 3 average isotope mass [classroom resources beanium isotopes](#)

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web the beanium lab activity aka isotopes and average atomic mass for elemental samples a mass spectrometer is used to measure the masses of each isotope as well as their relative abundance the results of these analyses is reported in the table of natural abundances chem ualberta ca massspec atomic mass abund pdf

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web 1 obtain a sample of beanium separate the pintonium navium and lentilium and then mass each group record the values in the data table 2 count the number of each isotope by counting the pintonium navium and lentilium 3 calculate the total number of isotopes

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web procedure obtain a sample of the classroom mixture of beanium sort your beanium into its three isotopic bean

types count the number of beans in each pile find the mass of each pile of beans determine the average mass of each type of bean based on the samples masses

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web good luck saftey 1 do not eat the beans 2 do not throw the beans 3 do not crush the beans don t mess with the beans pre lab questions 1 there are many beans in your container each bean represents an 2 there are three kinds of beans in your container each kind represents a different

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