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purpose what is an isotope what does it
mean to say that the atoms in a sample
of an element are isotopes of each
other ordinary beans are a lot bigger
than atoms but perhaps they can give
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have talked in class about isotopes
atoms of the same element that differ in
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lab questions define average atomic
mass write a mathematical equation
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isotope brownium beanium

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beanium isotope c whiteium of atoms
beans mass beans in cup mass empty
cup mass of atoms beans 2 minus 3
average mass of 1 bean 4 divided by 1
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calculate the average atomic mass of
beanium

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purpose this lab exercise is designed to
show you isotopes of an element in a
simulation form you will be asked to
gather data about the isotopes and
organize the data if atoms were as
large as beans they could be sorted
counted and massed

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web 1 obtain a sample of beanium separate the pintonium navium and lentilium and then mass each group record the values in the data table 2 count the number of each isotope by counting the pintonium navium and lentilium 3 calculate the total number of isotopes

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types count the number of beans in each pile find the mass of each pile of beans determine the average mass of each type of bean based on the samples masses

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