

Ha Come Emissario L Adda

Katrin Zwirgmaier

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you have almost enough race car parts

to build 100 cars except that you have

only 5 engines and 12 cylinders a if you

start building cars which component

engines or cylinders would limit the

number of

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web the substances that undergo

change are called reactants the new

substances are products sometimes

during a chemical reaction one type of

reactant will be used up before the

other reactants this reactant is the
 limiting reactant using the limiting
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limiting reactant calculating the

amount of product formed and

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chemists can optimize reaction

conditions and ensure efficient use of

resources in this article we have

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fill in the table below with the

maximum moles of water that can be

produced in each container q u indicate

which reactant limits the quantity of

water produced this is the limiting

reactant also show how much of the

other reactant the reactant in excess

will be left over

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web the limiting reagent concept allows us to calculate amounts of reactants used and products formed in a complete chemical reaction based on the stoichiometric relationships in a balanced chemical equation so far we have considered these relationships on the basis of masses and moles

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web step 4 the reactant that produces a smaller amount of product is the limiting reagent mg mg produces less mgo mgo than does o₂ o₂ 3 98 g mgo vs 25 2 g mgo therefore mg is the limiting reagent in this reaction step 5 the reactant that produces a larger

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 a determine the limiting reagent b
 determine the number of moles of h₂o
 h 2 o produced c determine the number

of grams of caso₄ c a s o 4 produced d
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 excess reagent left 1 make sure the
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