

S To Read As A Couple

Clemens Wendtner

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3 1 0 g 17 0 g

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 $47.867 \text{ g/mol} \times \text{moles Ti} = \text{mass Ti}$
 $\text{molar mass Ti} \times 4.12 \text{ mol Ti} = \text{mass Ti}$
 $47.867 \text{ g Ti} \times 1 \text{ mol Ti} = 197 \text{ g Ti}$ notag

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