

# Called Him Kitty

Michael Brown

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ions ending with ate such as chlorate

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carbonate co 3 2 and phosphate po 4 3  
the corresponding ite ion name has one  
less oxygen and the same charge

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formulas names and charges that  
cannot be modified in any way table 3 4  
1 lists the ion names and ion formulas  
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example no 3 is the nitrate ion it has  
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atoms and an overall 1 charge  
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or more covalently bonded atoms or a  
metal complex that acts like a single

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ions will not dissociate in water  
polyatomic ions can form compounds  
with each other

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than one atom they are called  
polyatomic ions polyatomic ions have  
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example the nitrate ion has one  
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with an overall charge of 1

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molecular ion is a covalent bonded set  
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some ions consist of groups of atoms bonded together and have an overall electric charge because these ions contain more than one atom they are called polyatomic ions polyatomic ions have characteristic formulas names and charges that should be memorized

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