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therefore $n = \frac{t}{t_{1/2}} \times n_0$ 100 now using the half life equation in example 1 we have

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web section learning objectives by the end of this section you will be able to do the following explain radioactive half life and its role in radiometric dating calculate radioactive half life and solve problems associated with radiometric dating section key terms half life and the rate of radioactive decay unstable nuclei decay

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web solution if we compare the time that has passed to the isotope's half life we note that 44 0 s is exactly 4 half lives so using the previous equation $n = \frac{t}{t_{1/2}} \times n_0$ 4 substituting and solving results in the following less than one third of a gram of 20 f remains skill building exercise the half life of 44 ti is 60 0 y

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