

Oh Happy Day By Sister Act

Ian Pickup

What is the product of the following reaction? 1)CH₃ OH - Socratic Dec 23, 2017 · These are ostensibly acid-base reactions.... For ammonium we could write.... $\text{NH}_4^+ + \text{HO}^- \rightleftharpoons \text{NH}_3(\text{aq}) + \text{H}_2\text{O}(\text{l})$ For methanol, the acid base reaction would proceed ... **Question #71ce2 - Socratic** Nov 9, 2014 · $\text{H}^+ + \text{OH}^- \rightarrow \text{H}_2\text{O}$ when the acid was added to the resulting solution. The H^+ and OH^- react in a 1:1 ratio. This tells us that the number of moles of H^+ used will be equal ... **Question #22e51 - Socratic** Products formed are; $\text{Cu}(\text{OH})_2 + \text{HCl} \rightarrow \text{CuCl}_2 + \text{H}_2\text{O}$ Since CuCl_2 is soluble in water.. But; $\text{Cu}(\text{OH})_2 + \text{HCl} \rightarrow \text{CuCl}_2 + \text{H}_2\text{O}$ " [Not Balanced]" Balancing the equation.. **Question #18488 - Socratic** The degree of dissociation α ($\alpha=0.0158$) $K_b=2.51 \times 10^{-6}$ color (white) (x)"mol/l") Triethylamine is a weak base and ionises: $(\text{CH}_3)_3\text{N} + \text{H}_2\text{O} \rightleftharpoons \dots$ **What is the chemical formula for lithium hydroxide? - Socratic** Apr 15, 2017 · Lithium is a Group 1 metal and commonly forms a M^+ ion. Hydroxide anion, OH^- , has a unit negative charge. When they make music together, there is thus 1:1 stoichiometry ... **Question #44ec4 - Socratic** That's a silly way to say carbon dioxide! $\text{CO}_2(\text{g}) + \text{Ca}(\text{OH})_2(\text{aq}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + \text{H}_2\text{CO}_3(\text{aq})$ May be wrong, but it seems reasonable. The gas would interact ... **What are the spectator ions in the reaction between sodium** Oct 16, 2016 · The nitrate and the sodium ions. $\text{Na}_2\text{CO}_3(\text{aq}) + 2\text{AgNO}_3(\text{aq}) \rightleftharpoons \text{Ag}_2\text{CO}_3(\text{s}) + 2\text{NaNO}_3(\text{aq})$ The net ionic equation is: $2\text{Ag}^+(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) \rightleftharpoons \dots$ Could a buffered solution be made by mixing aqueous ... - Socratic Mar 25, 2017 · Could a buffered solution be made by mixing aqueous solutions of HCl and NaOH ? Explain. Why isn't a mixture of a strong acid and its conjugate base considered a buffered ... *In each of the following pairs, which one has the lower viscosity?* Jul 31, 2015 · $\text{C}_2\text{H}_5\text{Cl}$ has weak dipole-dipole forces from the C-Cl bonds, but $\text{CH}_3\text{-OH}$ has strong hydrogen bonds. Confirmation: The viscosity of $\text{C}_2\text{H}_5\text{Cl}$ is 0.279 mPa·s, while the viscosity of ... **Question #627ef - Socratic** functional groups like -OH , -CHO , -CO- are present in carbohydrates. commonly two functional are present in a carbohydrate, -OH and -CHO , or -OH and -CO- . Aldose shows different reactions ...

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