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**M Woodhall**

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 powder 4 g was mixed with 25 ml of clear and colorless  
 diethyl ether solvent and the brown cloudy mixture was  
 filtered by gravity filtration with fluted fast flow filter paper  
 in a powder funnel into a 250 ml round bottom flask rbf  
 leaving light brown residue and beige filtrate diethyl ether  
 was

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web trimyristin in the major triglyceride found in nutmeg  
 representing 20 25 of the dried weight of ground nutmeg to  
 extract the trimyristin we performed a solid liquid extraction  
 using the organic solvent methylene chloride  $\text{CH}_2\text{Cl}_2$   
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web isolation of trimyristin from nutmeg purpose the  
 purpose of this experiment is to isolate myristic acid from  
 ground nutmeg seeds nutmeg seeds are extracted using  
 tbme to yield crude trimyristin then recrystallized to yield  
 pure trimyristin which will be hydrolyzed and undergo  
 esterification

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web read standard reflux on page 201 of the organic chem  
 lab survival manual prelab question 1 trimyristin is a  
 triglyceride what are triglycerides prelab question 2 how do

microwaves heat microwave procedure add approximately 25 g of ground nutmeg record exact mass and 15 ml of diethyl ether to microwave vessel

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web in this week s lab experiment the purpose of this lab is to new techniques isolate the chemical trimyristin from the common spice nutmeg this lab requires five techniques simple distillation vacuum filtration melting point determination reflux and extraction

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web aug 4 2008 the purpose of this experiment was extract isolate and purify the natural product trimyristin from the spice nutmeg the trimyristin was extracted by using the solvent diethyl ether and

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web trimyristin is found in the fixed oil of nutmeg the fixed

oil comprises approximately 24 40 of the nutmeg seed  
 trimyristin comprises 73 of the fixed oil overall trimyristin  
 should have percent recovery of 18 29 1 figure 1 shows how  
 trimyristin is triester formed from the dehydration reaction  
 between glycerol and myristic acid oh

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web purpose the purpose of this lab is to extract trimyristin  
 from nutmeg as well as synthesizing myristic acid from the  
 extracted trimyristin using base hydrolysis and naoh  
 acidification using hcl is necessary to characterize the  
 product by finding the melting point

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web in this lab trimyristin is extracted from ground nutmeg  
 occurring in many vegetable oils as well as myristic acid  
 trimyristin consists of long saturated hydrocarbon chains and  
 it is relatively nonpolar thus using a relatively nonpolar  
 solvent trimyristin can be easily extracted by the concept of  
 like dissolves like

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web the desired components of solids are dissolved in a  
 solvent then the remaining solid is filtered so that the  
 selected element and solvent are left over in this lab the  
 trimyristin was dissolved into the solvent and the remainder  
 of the nutmeg was filtered out

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 a type of triglyceride containing the fatty acid myristic acid  
 also known as tetradecanoic acid

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 nutmeg 2 purpose the purpose of this experiment is to isolate  
 a chemical in nutmeg called trimyristin through extraction  
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