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Julia Schneider

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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1823 studied the effect of temperature on the volume of a
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standard temperature and pressure stp for reporting
properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp
one mole of an ideal gas has a volume of about 22 4 l this is
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experiment the goal of the experiment was met the goal of
declaring what type of relationship it is was clear using the
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represented as a function of pressure and temperature as
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